

## ACID-BASE BALANCE

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- Close regulation of pH is necessary for cellular enzymes and other metabolic processes, which function optimally at the normal pH.
- Chronic, mild derangements in acid-base status may interfere with normal growth and development,
- Acute, severe changes in pH can be fatal.
- Control of acid-base balance depends on the kidneys, the lungs, and intracellular and extracellular buffers.

### Henderson-Hasselbach equation

- $\text{pH} = \text{pK} + \log \left( \frac{\text{HCO}_3^-}{\text{H}_2\text{CO}_3} \right)$   
 $\text{pK} = 6.1$  for  $\text{H}_2\text{CO}_3$   
 Normal ratio of  $\text{HCO}_3^-/\text{H}_2\text{CO}_3$   
 $= 20/1$   
 $\log 20 = 1.3$   
 Normal  $\text{pH} = 6.7 + 1.3 = 7.4$

### Physiological Buffers

- Bicarbonate
- Non-bicarbonate

### Bicarbonate buffer system

- Based on the relationship between carbon dioxide ( $\text{CO}_2$ ) and bicarbonate ( $\text{HCO}_3^-$ ):  
 $\text{CO}_2 + \text{H}_2\text{O} \leftrightarrow \text{H}^+ + \text{HCO}_3^-$

### Non-bicarbonate buffers

- Protein buffers
- Phosphate
- Bone

- **Protein buffers –**  
extracellular proteins - mostly albumin and intracellular proteins -including hemoglobin.
- **Phosphate buffers:**  
Phosphate can bind up to 3 hydrogen molecules, so it can exist as  $\text{PO}_4^{3-}$ ,  $\text{HPO}_4^{2-}$ , or  $\text{H}_2\text{PO}_4^-$ . However, at a physiologic pH, most phosphate exists as either  $\text{HPO}_4^{2-}$  or  $\text{H}_2\text{PO}_4^{1-}$ .  $\text{H}_2\text{PO}_4^{1-}$  is an acid, and  $\text{HPO}_4^{2-}$  is its conjugate base:  
 $\text{H}_2\text{PO}_4^{1-} \leftrightarrow \text{H}^+ + \text{HPO}_4^{2-}$

- **Bone** is an important buffer.
- Bone is basic—it is composed of compounds such as sodium bicarbonate and calcium carbonate—and thus, dissolution of bone releases base.
- This can buffer an acid load, although at the expense of bone density, if this occurs over an extended period.
- In contrast, bone formation, by consuming base, helps to buffer excess base.

**NORMAL ACID-BASE BALANCE**

- The **lungs** and kidneys maintain a normal acid-base balance.
- Carbon dioxide generated during normal metabolism is a weak acid.
- The lungs prevent an increase in the partial pressure of  $\text{CO}_2$  ( $\text{Pco}_2$ ) in the blood by excreting the  $\text{CO}_2$  that the body produces.
- $\text{CO}_2$  production varies depending on the body's metabolic needs, increasing with physical activity.
- The rapid pulmonary response to changes in the  $\text{CO}_2$  concentration occurs via central sensing of the  $\text{Pco}_2$  and a subsequent increase or decrease in ventilation to maintain a normal  $\text{Pco}_2$  (35–45 mm Hg).
- An increase in ventilation decreases the  $\text{Pco}_2$ , and a decrease in ventilation increases the  $\text{Pco}_2$ .

- The **kidneys** excrete endogenous acid.
- The kidneys regulate the serum bicarbonate concentration by modifying acid excretion in the urine. This requires a 2-step process.
- First, the renal tubules resorb the bicarbonate that is filtered at the glomerulus.
- Second, there is tubular secretion of hydrogen ions.
- The urinary excretion of hydrogen ions generates bicarbonate that neutralizes endogenous acid production

**Normal ABG values**

- pH            7.35 – 7.45
- $\text{pCO}_2$         35 – 45 mm Hg
- $\text{pO}_2$          50 – 70 mm Hg
- $\text{HCO}_3$        20 – 24 meq/L
- BE             $\pm 5$

**Terminology**

- Acidosis         $\text{pH} < 7.3$
- Alkalosis        $\text{pH} > 7.5$
- Hypercapnia     $\text{paCO}_2 > 50 \text{ mmHg}$
- Hypocapnia     $\text{paCO}_2 < 30 \text{ mmHg}$
- Hypoxia         $\text{pO}_2 < 50 \text{ mm Hg}$
- Hyperoxia       $\text{pO}_2 > 70 \text{ mm Hg}$

- The **plasma anion gap** is useful for evaluating patients with a metabolic acidosis.
- Anion gap =  $[Na^+] - [Cl^-] - [HCO_3^-]$
- A normal anion gap is 4–11.
- The anion gap is the difference between the measured cation (sodium) and the measured anions (chloride + bicarbonate).
- The anion gap is also the difference between the unmeasured cations (potassium, magnesium, calcium) and the unmeasured anions (albumin, phosphate, urate, sulfate).
- An increased anion gap occurs when there is an increase in unmeasured anions.

- ### Calculated values
- Buffer Base (BB)  
48-50 mmol/L  
Hb(25%),HCO3(50%),Proteins(25%)
  - Base Excess( ABE , SBE)  
 $\pm 5$  mmol/L

### Simple disorder

Disorder	pH	pCO2	HCO3
Metabolic acidosis	↓	↓	↓
Metabolic alkalosis	↑	↑	↑
Resp. acidosis	↓	↑	↑
Resp. alkalosis	↑	↓	↓

- ### Compensatory mechanisms
- Respiratory  
Renal effect on bicarbonate
  - Metabolic  
Respiratory effect on CO2

- ### Metabolic Acidosis
- There are many causes of a metabolic acidosis which occur via 3 basic mechanisms:
- Loss of bicarbonate from the body
  - Impaired ability to excrete acid by the kidney
  - Addition of acid to the body (exogenous or endogenous)

- ### Causes of Metabolic Acidosis
- **NORMAL ANION GAP**
    - Diarrhea
    - Renal tubular acidosis (RTA)
    - Distal (type I) RTA (MIM 179800/602722/267300)<sup>[\*]</sup>
    - Proximal (type II) RTA (MIM 604278)<sup>[†]</sup>
    - Hyperkalemic (type IV) RTA (MIM 201910/264350/177735/145260)<sup>[‡]</sup>
    - Urinary tract diversions
    - Posthypocapnia
    - Ammonium chloride intake

- **INCREASED ANION GAP**

- Lactic acidosis
  - Tissue hypoxia (shock, hypoxemia, severe anemia)
  - Liver failure
  - Malignancy
  - Intestinal bacterial overgrowth
  - Inborn errors of metabolism
  - Medications (nucleoside analogues, metformin)

- Ketoacidosis
  - Diabetic
  - Starvation
  - Alcoholic
- Kidney failure
- Poisoning
  - Ethylene glycol
  - Methanol
  - Salicylate
  - Paraldehyde
  - Toluene
- Inborn errors of metabolism

### Causes of Metabolic Alkalosis

- **CHLORIDE RESPONSIVE (URINARY CHLORIDE < 15 mEq/L)**
  - Gastric losses (emesis or nasogastric suction)
  - Diuretics (loop or thiazide)
  - Chloride-losing diarrhea (MIM 214700)
  - Chloride-deficient formula
  - Cystic fibrosis (MIM 219700)
  - Post-hypercapnia

### CHLORIDE RESISTANT (URINARY CHLORIDE > 20 mEq/L)

- High blood pressure
  - Adrenal adenoma or hyperplasia
  - Glucocorticoid-remedial aldosteronism (MIM 103900)
  - 17 $\alpha$ -hydroxylase deficiency (MIM 202110)
  - Renovascular diseaseRenin-secreting tumor
  - 11 $\beta$ -hydroxylase deficiency (MIM 202010)
  - Cushing syndrome
  - 11 $\beta$ -hydroxysteroid dehydrogenase deficiency (MIM 218030)
  - Licorice ingestion
  - Liddle syndrome (MIM 177200)
- Normal blood pressure
  - Gitelman syndrome (MIM 263800)
  - Bartter syndrome (MIM 602023/607364/602522/241200/601678)
  - Autosomal dominant hypoparathyroidism (MIM 146200)
  - Base administration

### Causes of respiratory alkalosis

- **HYPOXEMIA OR TISSUE HYPOXIA**
  - Pneumonia
  - Pulmonary edema
  - Cyanotic heart disease
  - Congestive heart failure
  - Asthma
  - Severe anemia
  - High altitude
  - Laryngospasm
  - Aspiration
  - Carbon monoxide poisoning
  - Pulmonary embolism
  - Interstitial lung disease
  - Hypotension

- **LUNG RECEPTOR STIMULATION**

- Pneumonia
- Pulmonary edema
- Asthma
- Pulmonary embolism
- Hemothorax
- Pneumothorax
- Respiratory distress syndrome (adult or infant)

- **CENTRAL STIMULATION**

- Central nervous system disease
  - Subarachnoid hemorrhage
  - Encephalitis or meningitis
  - Trauma
  - Brain tumor
  - Stroke
- Fever
- Pain
- Anxiety (panic attack)
- Psychogenic hyperventilation or anxiety
- Liver failure
- Sepsis
- Pregnancy
- Medications
  - Salicylate intoxication
  - Theophylline
  - Progesterone
  - Exogenous catecholamines
  - Caffeine
- Mechanical ventilation
- Hyperammonemia
- Extracorporeal membrane oxygenation or hemodialysis

## Causes of respiratory acidosis

- **CENTRAL NERVOUS SYSTEM DEPRESSION**

- Encephalitis
- Head trauma
- Brain tumor
- Central sleep apnea
- Primary pulmonary hypoventilation (Ondine curse)
- Stroke
- Hypoxic brain damage
- Obesity-hypoventilation (Pickwickian syndrome)
- Increased intracranial pressure
- Medications
  - Narcotic
  - Barbiturates
  - Anesthesia
  - Benzodiazepines
  - Propofol
  - Alcohols

## DISORDERS OF THE SPINAL CORD, PERIPHERAL NERVES, OR NEUROMUSCULAR JUNCTION

- Diaphragmatic paralysis
- Guillain-Barré syndrome
- Poliomyelitis
- Spinal muscular atrophies
- Tick paralysis
- Botulism
- Myasthenia
- Multiple sclerosis
- Spinal cord injury
- Medications
  - Vecuronium
  - Aminoglycosides
  - Organophosphates (pesticides)

## RESPIRATORY MUSCLE WEAKNESS

- Muscular dystrophy
- Hypothyroidism
- Malnutrition
- Hypokalemia
- Hypophosphatemia
- Medications
- Succinylcholine
- Corticosteroids

- **PULMONARY DISEASE**

- Pneumonia
- Pneumothorax
- Asthma
- Bronchiolitis
- Pulmonary edema
- Pulmonary hemorrhage
- Adult respiratory distress syndrome
- Respiratory distress syndrome, neonatal
- Cystic fibrosis
- Bronchopulmonary dysplasia
- Hypoplastic lungs
- Meconium aspiration
- Pulmonary thromboembolus
- Interstitial fibrosis

- **UPPER AIRWAY DISEASE**

- Aspiration
- Laryngospasm
- Angioedema
- Obstructive sleep apnea
- Tonsillar hypertrophy
- Vocal cord paralysis
- Extrinsic tumor
- Extrinsic or intrinsic hemangioma

- **MISCELLANEOUS**

- Flail chest
- Cardiac arrest
- Kyphoscoliosis
- Decreased diaphragmatic movement due to ascites or peritoneal dialysis